FACULTY OF PHARMACY

B. Pharmacy III-Semester (PCI) (Main & Backlog) Examination, January 2020

Subject: Physical Pharmaceutics - I

Time: 3 Hours Max. Marks: 75

Note: Answer all Questions from Part – A, and Two questions from Part – B, and any Seven questions from Part – C.

PART - A (10 X 2 = 20)

- 1. What is sorensen's pH scale?
- 2. What is buffer? Write the buffer equation.
- 3. What are solid dispersions?
- 4. What is common ion effect? Explain.
- 5. What is Refractive index?
- 6. What are ampholytes, Give examples?
- 7. Write the solubility of drug as part of solvent required for a part of solute as per USP.
- 8. Define complexation & chelation.
- 9. Define Detergency with example.
- 10. Define optical activity and specific rotation.

$PART - B (2 \times 10 = 20)$

- 11. State and explain the relative lowering of vapour pressure of Roult's law. Explain its limitations.
- 12. What is Polymorphism? Give 4 examples of drugs exhibiting Polymorphism, Write its significance.
- 13. Explain in detail methods of adjustment of tonicity.

$$PART - C (7 \times 5 = 35)$$

- 14. Write a note on Liquid Crystalls.
- 15. Write a short note on -
 - (a) Noyes-whitney equation (b) Dankwert's Model
- 16. State distribution law. Discuss the applications.
- 17. Explain about Protein binding.
- 18. Define refractive index. Describe snell's law in detail.
- 19. Describe capillary rise method to determine surface tension of liquid.
- 20. Define complexation. What are types of complexes? Write about inclusion complex.
- 21. Enlist various methods of liquefaction gases. Explain any two.
- 22. Explain the difference between ideal solution and real solution.
